

Pressure Swing Adsorption as an Efficient Tool for the Separation of Carbon Dioxide from Flue Gases

**K.Warmuzinski, M.Tanczyk, M.Jaschik, A.Wojdyla
Institute of Chemical Engineering
Polish Academy of Sciences
Gliwice**



1. Introduction

- **fossil fuels – the CO₂ problem**
- **abatement options**

2. Pressure swing adsorption

- **basic principles**
- **equilibria**
- **kinetics**

3. Scope of the study

- **adsorbent selection**
- **simulations**

4. Experimental

- **equilibria**
- **adsorbent selection**
- **kinetics**
- **PSA separation experiments**

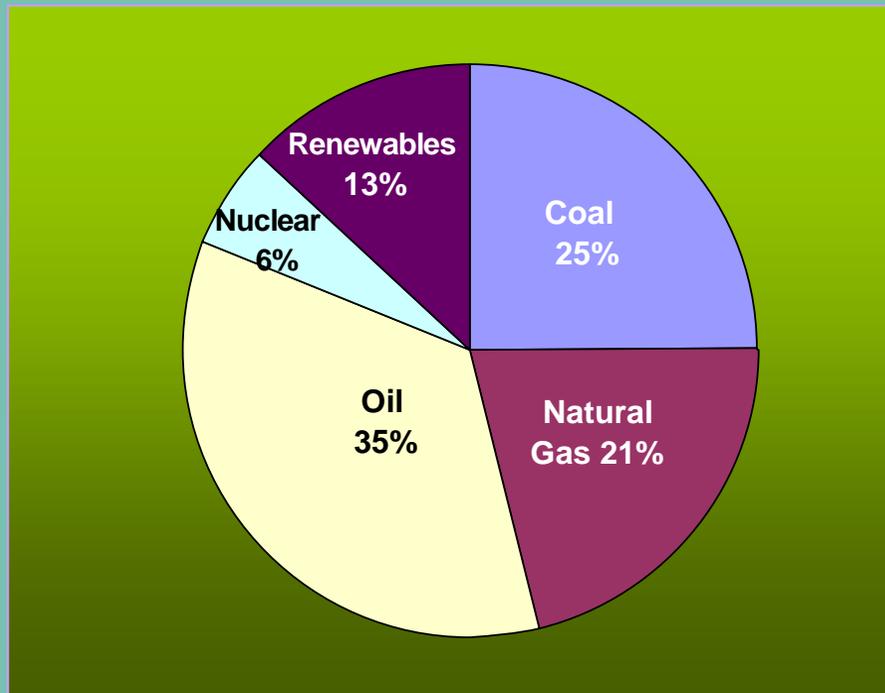
5. Mathematical modelling

6. Development of an efficient PSA cycle

- **CO₂ purity**
- **CO₂ recovery**

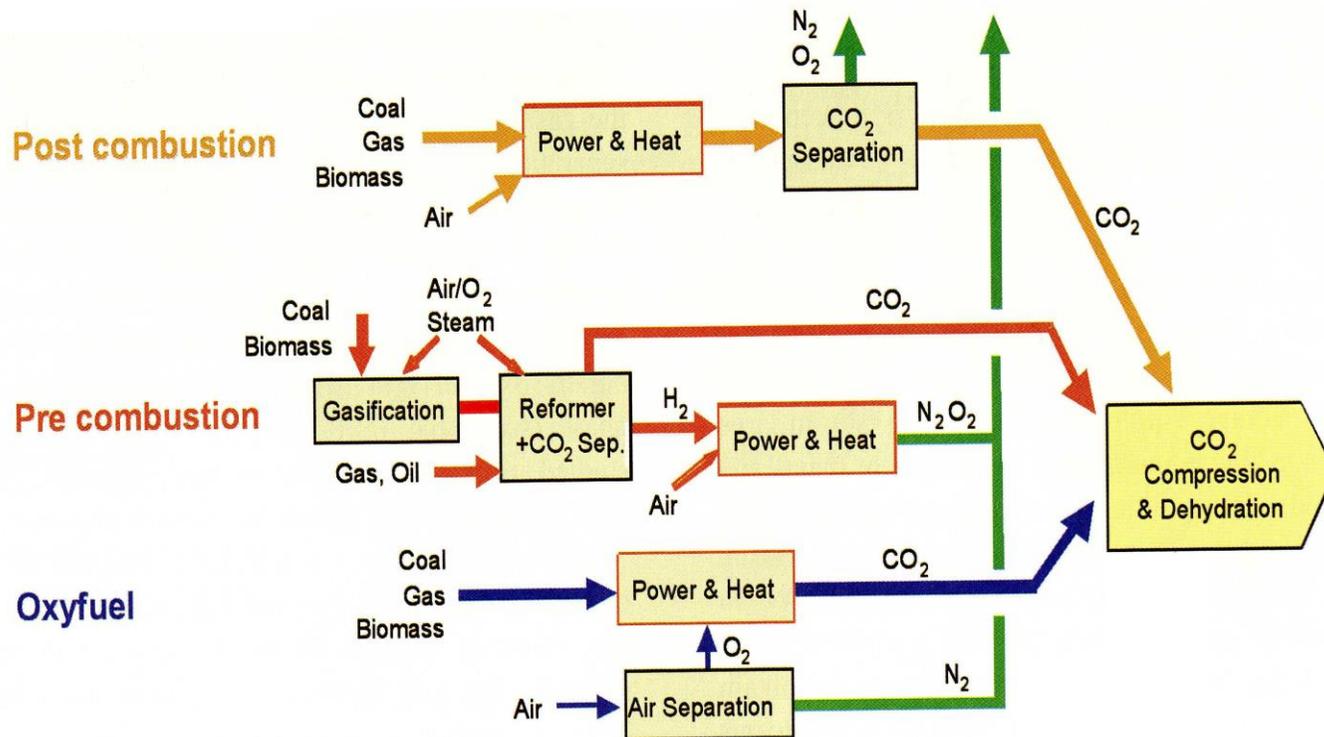
7. Conclusions

Fossil fuels provide over 80% of primary energy

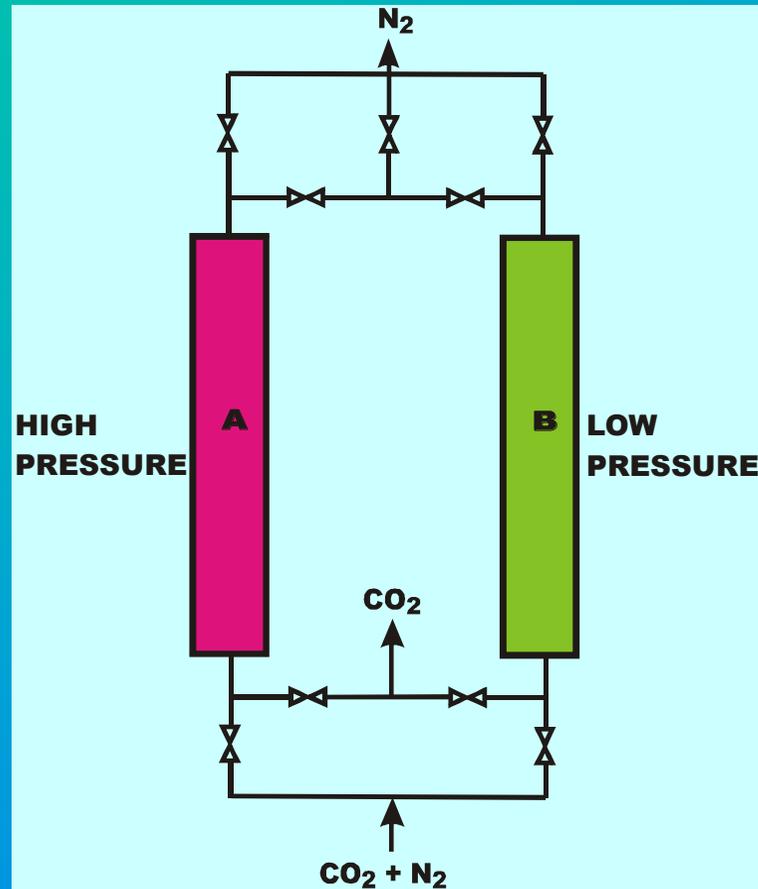


Today, almost 30 billion tonnes of CO₂ are emitted every year as a result of using fossil fuels

Carbon dioxide capture systems



The principle of pressure swing adsorption





Adsorption equilibrium

- how **much** of a given species can be adsorbed

Adsorption kinetics

- how **fast** a species is being adsorbed



Objectives of the present study

- **to select the best adsorbent for the separation of CO₂/N₂ mixtures (based on experiments)**
- **to develop an efficient PSA cycle for the separation of CO₂/N₂ mixtures (based on experiment, modelling and simulations)**



Adsorbents studied

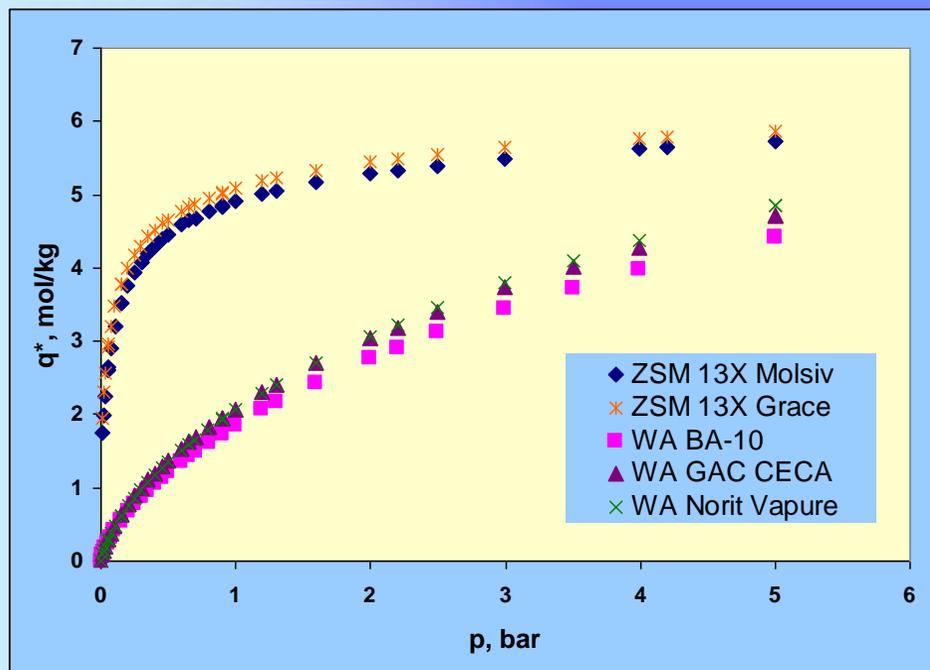
- **BA-10**
- **GAC CECA**
- **Norit Vapure**

activated carbons

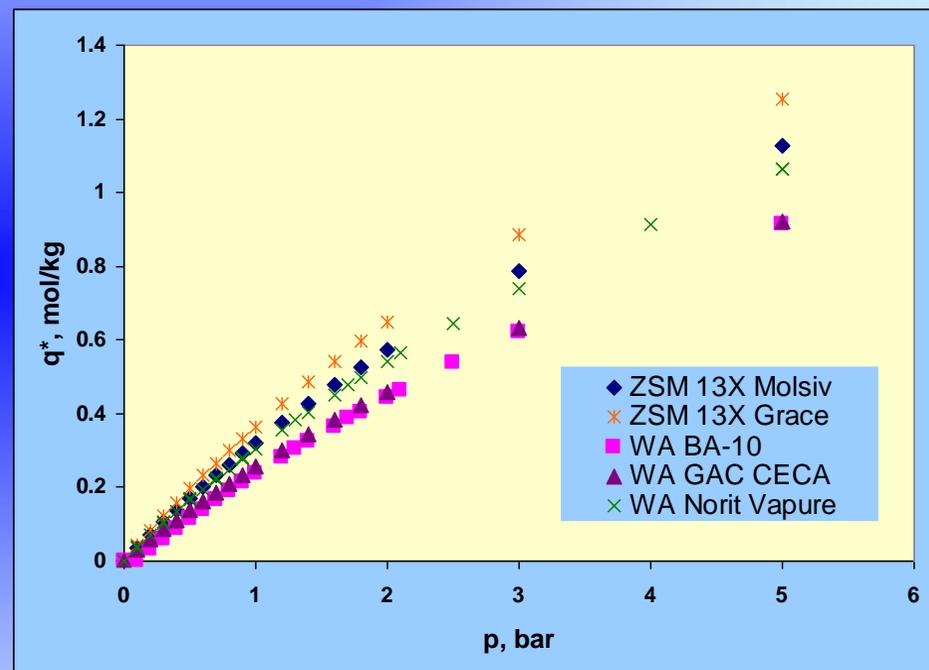
- **Molsiv 13X**
- **Grace 13X**

zeolite molecular sieves

Representative adsorption isotherms for CO_2 and N_2 at 20°C



CARBON DIOXIDE



NITROGEN

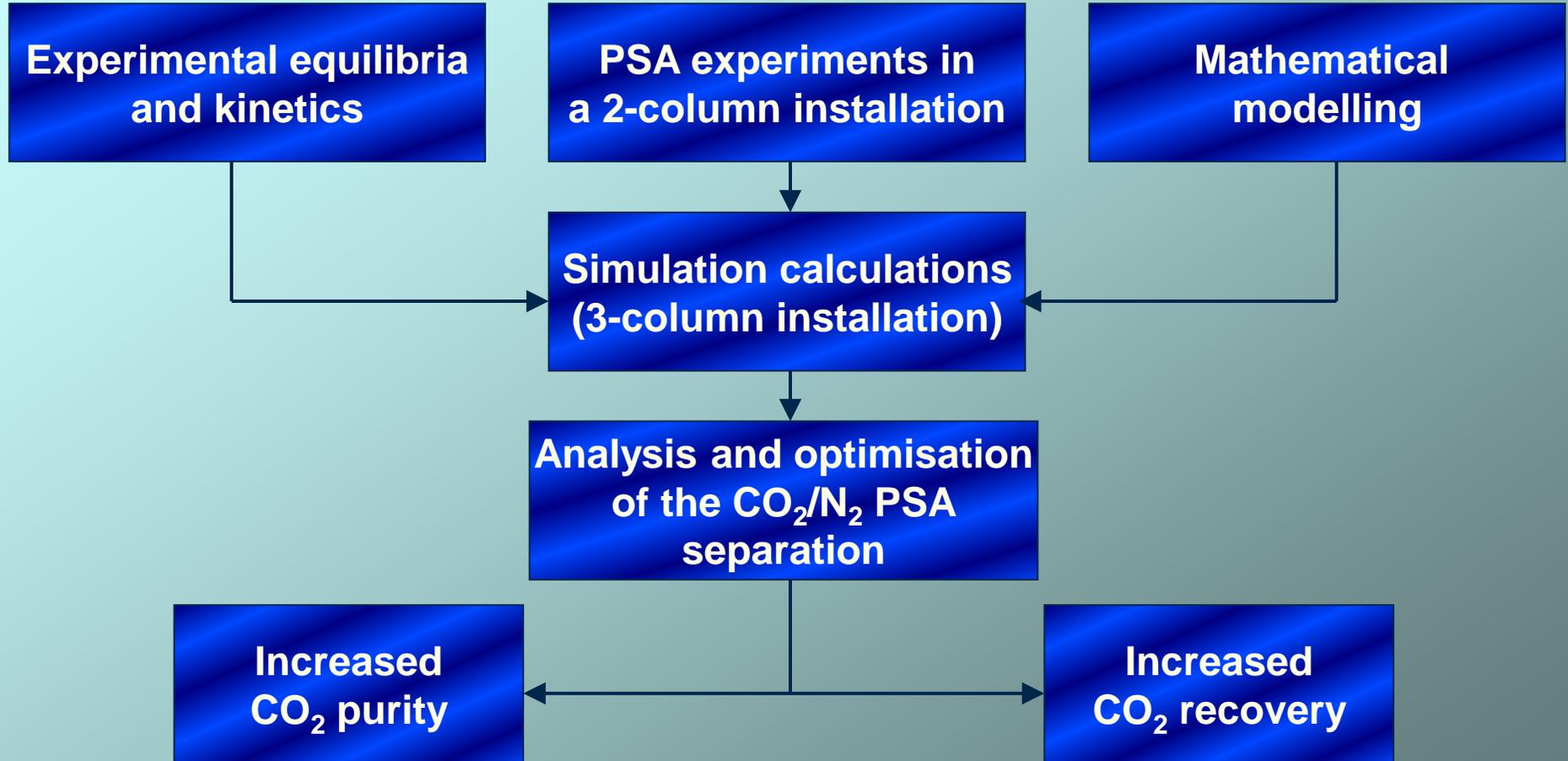
Adsorbent selected: ZMS 13X (Grace)

**Kinetic constants for CO₂ and N₂ over
the adsorbent selected**

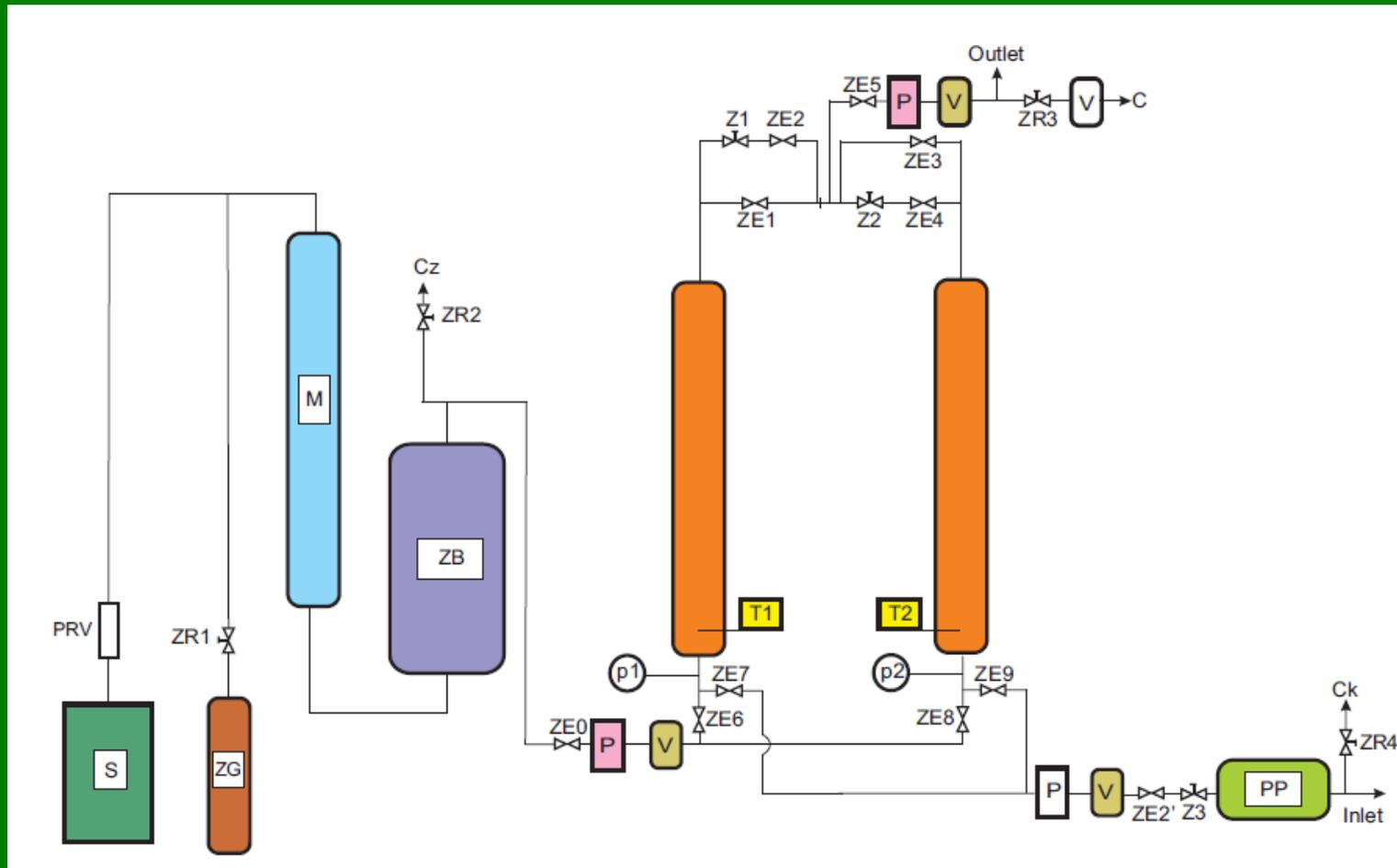
Gas	D/r ² , s ⁻¹			
	20°C	40°C	60°C	80°C
CO ₂	9.06 · 10 ⁻⁴	1.07 · 10 ⁻³	1.33 · 10 ⁻³	1.59 · 10 ⁻³
N ₂	4.08 · 10 ⁻³	4.79 · 10 ⁻³	4.65 · 10 ⁻³	4.86 · 10 ⁻³



Development of an efficient PSA cycle



Experimental PSA installation





Basic assumptions of the model

- **thermal equilibrium between the gas and the solid phase**
- **plug flow with axial dispersion**
- **negligible pressure drop over the adsorbent**
- **the fluid phase is an ideal gas**

PSA cycle in the 3-column installation

Column \ Stage	1	2	3
1	F↑	D↑ Pu↑	D↓ R↓ P↓
2	D↓ R↓ P↓	F↑	D↑ Pu↑
3	D↑ Pu↑	D↓ R↓ P↓	F↑

F – feed, D – depressurisation, P – pressurisation

Pu – purge, R – vacuum regeneration

↑ – cocurrent flow, ↓ – countercurrent flow



Feed

- **15% CO₂, 85% N₂**

Improved process conditions

- **optimised flow rates during purge and vacuum regeneration**

Results

- **CO₂ recovery close to 100%**
- **CO₂ purity in excess of 80%**



Conclusions

The improved PSA process can enrich the 15/85 CO₂/N₂ mixture to a level of over 80% of CO₂, with an almost complete recovery of this species. Such efficiency can be achieved by a suitable selection of flow rates during purge with the enriched gas and vacuum regeneration.

Thank you